2002 Vol. 4, No. 15 2469 - 2472

Asymmetric Synthesis of (–)-Adaline

Toshimasa Itoh, Naoki Yamazaki, and Chihiro Kibayashi*

School of Pharmacy, Tokyo University of Pharmacy and Life Science, 1432-1 Horinouchi, Hachioji, Tokyo 192-0392, Japan kibayasi@ps.toyaku.ac.jp

Received April 18, 2002

ABSTRAC1

An enantioselective total synthesis of (-)-adaline has been achieved starting from a chiral 6,6-disubstituted piperidone derivative previously prepared by diastereoselective allylation of a chiral tricyclic N-acyl-N,O-acetal. The key steps include lithium ion-activated S_N2-type alkynylation of the tricyclic N,O-acetal leading to exclusive formation of the (6S)-ethynylpiperidine and ring-closing olefin metathesis of the (2R,6S)-cis-2,6-dialkenylpiperidine for constructing the bridged azabicyclononane.

We previously reported TiCl₄-mediated asymmetric allylation employing tricyclic N-acyl-N,O-acetals, in which enantiomeric 2-(1-aminoethyl)phenol (3) has been proven useful as a chiral auxiliary. In connection with this investigation, most recently we have developed a convenient practical asymmetric synthesis of 3 via nucleophilic addition of methyllithium to a chiral oxime ether.² We have also demonstrated that this asymmetric allylation protocol is efficiently applicable to the total synthesis of various alkaloids, and the first enantioselective synthesis of a new coccinellid alkaloid (-)-adalinine (1)3,4 was achieved on the basis of a diastereoselective addition of allylsilane to an in situ generated

N-acyliminium ion 6 to form the (R)-allylated piperidone 7

as shown in Scheme 1.1a We envisioned compound 7 utilized as the key synthetic intermediate in this scheme could be

Scheme 1 Me OH 3 TiCl₄ SiMe₃

^{(1) (}a) Yamazaki, N.; Ito, T.; Kibayashi, C. Tetrahedron Lett. 1999, 40, 739-742. (b) Yamazaki, N.; Ito, T.; Kibayashi, C. Org. Lett. 2000, 2, 465-467. (c) Yamazaki, N.; Dokoshi, W.; Kibayashi, C. Org. Lett. 2001, 3, 193-

applicable to the synthesis of (—)-adaline (2), $^{4-6}$ which is the major defensive alkaloid from the European ladybug, *Adalia bipunctata*, and is suggested⁷ to be biogenetically related to (—)-adalinine (1) isolated as a minor defensive alkaloid. In this paper, we describe the enantioselective total synthesis of (—)-adaline (2)⁸ starting from the (6*S*)-piperidone 7 by an S_N2 -like alkynylation using a tricyclic *N*,*O*-acetal as an extension of the previous *N*-acyl-*N*,*O*-acetal-based allylation, followed by ring-closing olefin metathesis.

Reductive lactam ring opening of 7 was carried out by employing LiH₂NBH₃⁹ (THF at 40 °C) to give the amino alcohol 8 in 88% yield (Scheme 2). This compound was cyclized by a one-pot procedure with TPAP-NMO in acetonitrile (4A MS, rt, 1 h) that involved oxidation to the aldehyde 9 followed by dehydrocondensation to afford the tricyclic N,O-acetal 10 (80% yield) as a single stereoisomer, where the chiral auxiliary 3 was incorporated. The 5aS configuration of 10 was confirmed by the NOESY interaction between 5a-H and the methyl group at C11 as depicted in Figure 1. Upon treatment of 10 using lithium acetylide ethylenediamine complex (HC≡CLi•H₂NCH₂CH₂NH₂)¹⁰ in THF at 40 °C, the nucleophilic alkynylation¹¹ proceeded with complete inversion of configuration at the reaction center and, unexpectedly, with concomitant removal of the 1-(2hydroxyphenyl)ethyl function via C-N bond cleavage, leading to the (6S)-ethynylpiperidine 12 (88%) as a single diastereomer along with racemic 1-(2-hydroxyphenyl)ethanol (14) (73%). The S absolute stereochemistry at C6 of 12 was determined by the NOESY experiment, which revealed the equatorial orientation of the C6-ethynyl group, syn to the C2-allyl group as shown in Figure 2. The exclusive formation of the (6S)-isomer 12 is a consequence of the coordination of the lithium ion to the oxygen of the N,Oacetal, activating the C-O bond for nucleophilic attack of the lithium acetylide to give the S_N2-type ring opening of the N,O-acetal.¹³ In the second step, the generated lithium

Scheme 2^a

^a Key: (a) LiH₂NBH₃, THF, 40 °C; (b) TPAP, NMO, MeCN, 4A MS, rt; (c) HC≡CLi•H₂NCH₂CH₂NH₂ (5 equiv), THF, 40 °C.

phenoxide 11 may form a tight six-membered chelate complex, facilitating the cleavage of the benzylic carbon—

Figure 1. Selected NOESY correlation for 10.

2470 Org. Lett., Vol. 4, No. 15, 2002

⁽²⁾ Yamazaki, N.; Atobe, M.; Kibayashi, C. *Tetrahedron Lett.* **2001**, *42*, 5029–5032.

⁽³⁾ For isolation and structure elucidation, see: Lognay, G.; Hemptinne, J. L.; Chan, F. Y.; Gaspar, C. H.; Marlier, M.; Braekman, J. C.; Daloze, D.; Pasteels, J. M. *J. Nat. Prod.* **1996**, *59*, 510–511.

⁽⁴⁾ For a recent review of the coccinellid alkaloids, see: King, A. G.; Meinwald, J. Chem. Rev. 1996, 96, 1105–1122.

⁽⁵⁾ For the isolation and structural determination of adaline, see: Tursch, B.; Braekman, J. C.; Daloze, D.; Hootele, C.; Losman, D.; Karlsson, R.; Pasteels, J. M. *Tetrahedron Lett.* **1973**, 201–202.

⁽⁶⁾ For the determination of the absolute configuration of natural (–)-adaline, see: Tursch, B.; Chome, C.; Braekman, J. C.; Daloze, D. *Bull. Soc. Chim. Belg.* **1973**, 82, 699–703.

⁽⁷⁾ Laurent, P.; Lebrun, B.; Braekman, J.-C.; Daloze, D.; Pasteels, J. M. *Tetrahedron* **2001**, *57*, 3403–3412.

⁽⁸⁾ For the synthesis of (−)-adaline, see: (a) Hill, R. K.; Renbaum, L. A. *Tetrahedron* **1982**, *38*, 1959−1963. (b) Yue, C.; Royer, J.; Husson, H.-P. *J. Org. Chem.* **1992**, *57*, 4211−4214. For the synthesis of (±)-adaline, see: (c) Alder, v. K.; Betzing, H.; Kuth, R.; Dortmann, H.-A. *Liebigs Ann. Chem.* **1959**, *620*, 73−87. (d) Gnecco Medina, D. H.; Grierson, D. S.; Husson, H.-P. *Tetrahedron Lett.* **1983**, *24*, 2099−2102. (e) Gössinger, E.; Witkop, B. *Monatsh. Chem.* **1980**, *111*, 803−811. (f) Davison, E. C.; Holmes, A. B.; Forbes, I. T. *Tetrahedron Lett.* **1995**, *36*, 9047−9050.

⁽⁹⁾ Myers, A. G.; Yang, B. H.; Kopecky, D. J. *Tetrahedron Lett.* **1996**, *37*, 3623–3626.

⁽¹⁰⁾ Beumel, O. F., Jr.; Harris, R. F. J. Org. Chem. 1964, 29, 1872–1976.

⁽¹¹⁾ Treatment of $\mathbf{7}$ with vinylmagnesium bromide or vinyllithium in THF resulted in no reaction.

⁽¹²⁾ Chiral HPLC (Chiralpak AD column) analysis of compound 14 showed two peaks with exactly equal integration for the both enantiomers.

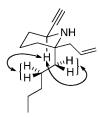


Figure 2. Selected NOE correlation for 12.

nitrogen bond to furnish a transient trienone **13**, which can be immediately aromatized in the presence of water to the hydroxyphenol **14** with complete racemization.¹²

For ring-closing metathesis (RCM),¹⁴ the enyne **12** was transformed to the *cis*-2,6-dialkenylpiperidine **15** by hydrogenation of the alkynyl moiety using Lindlar catalyst (Scheme 3). Since free bases are claimed to be ineffective

^a Key: (a) Lindlar catalyst, H₂, MeOH; (b) HCl/MeOH; (c) Grubbs catalyst **20** (0.25 equiv), benzene, 50 °C; (d) HCl/MeOH, then HC(OMe)₃, TsOH; (e) Grubbs catalyst **21** (0.15 equiv), benzene, 50 °C.

for RCM mediated by Grubbs ruthenium catalyst due to poisoning of the catalyst by the amine functionality, ¹⁵ **15** was converted to the hydrochloride salt and the RCM reaction was applied to the construction of the Δ^3 -homotro-

pane **16**. However, attempts to induce the reaction using Grubbs catalyst **20**¹⁶ failed, leading to recovery of the starting material. This failure is most probably due to diequatorial arrangement of the 2,6-dialkenyl substituents in **15** in a chair form as indicated by the NOESY correlation (Figure 3),

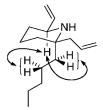
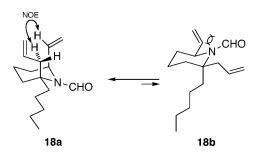


Figure 3. Selected NOE correlation for 15.

which is inadequately disposed for the requisite ring closing to 16. Indeed, it has been shown that RCM requires the substrate conformation constraining the olefins in close proximity.¹⁷ We thus envisaged as a substrate for the RCM the use of the N-acyl derivative of 14, wherein the 2,6dialkenyl substituents would prefer the diaxial orientation required for the ring closure to avoid unfavorable $A^{(1,3)}$ strain¹⁸ between the vinyl group at C6 and the *N*-acyl group. Thus, the hydrochloride salt of 12 was treated with trimethylorthoformate to give the formamide 17, which was converted to the cis-2,6-dialkenylformamide 18 by Lindlarcatalyzed partial hydrogenation. The NOESY data revealed that 18 preferentially exists in the expected conformation **18a** having the diaxial disposition of the dialkenyl substituents. The RCM reaction applied to 18 smoothly proceeded with Grubbs catalyst 20 (25 mol %) in benzene at 50 °C for 6 h to give the Δ^3 -homotropane **19** in high yield (90%).¹⁹ When the second-generation Grubbs catalyst 21²⁰ (15 mol %) was employed, the reaction proceeded more efficiently and was completed in 40 min affording 19 in almost quantitative yield.



The resulting homotropane **19** was subjected to dihydroxylation with OsO₄—NMO to form the diol **22** as the only product (97%) owing to an exclusive convex approach of

Org. Lett., Vol. 4, No. 15, 2002

⁽¹³⁾ For a review on the reactions and synthetic application of bicyclic lactams containing *N*,*O*-acetals, see: Romo, D.; Meyers, A. I. *Tetrahedron* **1991**, *47*, 9503–9569.

⁽¹⁴⁾ For reviews, see: (a) Grubbs, R. H.; Miller, S. J.; Fu, G. C. *Acc. Chem. Res.* **1995**, 28, 446–552. (b) Grubbs, R. H.; Chang, S. *Tetrahedron* **1998**, 54, 4413–4450.

^{(15) (}a) Fu, G. C.; Nguyen, S. T.; Grubbs, R. H. J. Am. Chem. Soc. 1993, 115, 9856–9857. (b) Suzuki, H.; Yamazaki, N.; Kibayashi, C. Tetrahedron Lett. 2001, 42, 3013–3015.

⁽¹⁶⁾ Schwab, P.; Grubbs, R. H.; Ziller, J. W. J. Am. Chem. Soc. 1996, 118, 100–110.

^a Key: (a) OsO₄, NMO, MeCN-H₂O; (b) TBDMSOTf, Et₃N, CH₂Cl₂; (c) CS₂, NaH, MeI, THF; (d) AIBN, Bu₃SnH, benzene, reflux; (e) TBAF, THF; (f) Li·H₂NCH₂CH₂NH₂; (g) PCC, CH₂Cl₂.

26

25

the oxidant (Scheme 4). The configuration of the C3 and C4 stereogenic centers newly generated by the dihydroxylation was assigned on the basis of the coupling constants observed for the vicinal protons in the ¹H NMR spectra of compound 23 (see below). Regioselective protection of 22 was achieved with TBDMSOTf and Et₃N to give a separable

mixture of 3-siloxy alcohol 23 and 4-siloxy alcohol 24 with 10:1 chemoselectivity in a total yield of 98%. Conversion of 23 into 25 was accomplished by the Barton-McCombie radical-induced deoxygenation procedure.²¹ Thus, **23** was converted to the methyl xanthate (CS2, NaI, MeI), which was treated with AIBN and Bu₃SnH in refluxing benzene to give 25 in 82% yield. Removal of the TBDMS and formyl protecting groups in 25 was performed by treatment with TBAF in wet THF and Li-ethylenediamine complex,²² respectively, to furnish the amino alcohol (dihydroadaline) **26**, which underwent PCC oxidation to provide (-)-adaline (2), $[\alpha]^{28}_D$ -11.4 (c 0.7, CHCl₃) [lit.⁵ $[\alpha]^{20}_D$ -11 (c 2, CHCl₃)]. Spectral data (¹H NMR, ¹³C NMR, and MS) of the synthetic material were identical to those reported^{8b} for the natural adaline.

In conclusion, we have demonstrated the utility of 2-(1aminoethyl)phenol as a chiral auxiliary on the highly diastereoselective S_N 2-type alkynylation of the tricyclic N,Oacetal leading to exclusive formation of the (6S)-ethynylpiperidine. Using this reaction as well as ring-closing olefin metathesis for constructing the bridged azabicyclononane, the enantioselective total synthesis of (-)-adaline has been achieved.

Supporting Information Available: Experimental procedures and characterization data for all new compounds and adaline. This material is available free of charge via the Internet at http://pubs.acs.org.

OL0200807

(17) (a) Forbes, M. D. E.; Patton, J. T.; Myers, T. L.; Maynard, H. D.; Smith, D. W., Jr.; Schulz, G. R.; Wagener, K. B. J. Am. Chem. Soc. 1992, 114, 10978—10980. (b) Miller, S. J.; Kim, S.-H.; Chen, Z.-R.; Grubbs, R. H. J. Am. Chem. Soc. 1995, 117, 2108—2109. (c) Marsella, M. J.; Maynard, H. D.; Grubbs, R. H. Angew. Chem., Int. Ed. Engl. 1997, 36, 1101-1103.

(18) For reviews, see: (a) Johnson, F. Chem. Rev. 1968, 68, 375-413. (b) Hoffmann, R. W. Chem. Rev. 1989, 89, 1841-1860.

(19) After completion of our own work, a similar RCM approach to bridged azabicyclic structures was reported by Martin et al.; see: Neipp, C. E.; Martin, S. F. Tetrahedron Lett. 2002, 43, 1779-1782.

(20) Scholl, M.; Ding, S.; Lee, C. W.; Grubbs, R. H. Org. Lett. 1999, 1,

(21) Barton, D. H. R.; McCombie, S. W. J. Chem. Soc., Perkin Trans. 1 **1975**, 1574-1585.

(22) (a) Hallsworth, A. S.; Henbest, H. B. J. Chem. Soc. 1957, 4604-4608. (b) Brown, H. C.; Ikegami, S.; Kawakami, J. H. J. Org. Chem. 1970, 35, 3243-3245. (c) Mueller, R. H.; Thompson, M. E.; DiPardo, R. M. J. Org. Chem. 1984, 49, 2217-2231.

2472 Org. Lett., Vol. 4, No. 15, 2002